

# Desorption of volatile molecules from interstellar carbonaceous dust analogs

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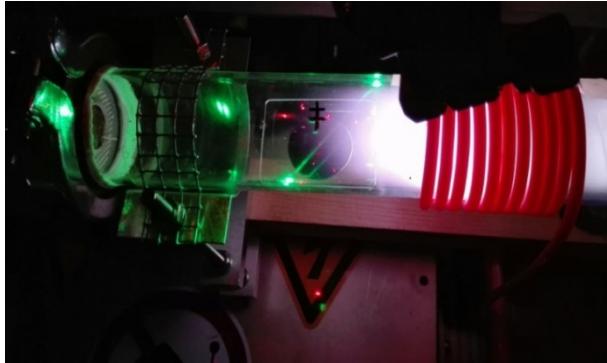
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# Outline

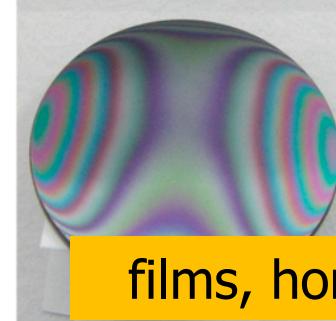
- Hydrogenated amorphous carbon as ISM dust analogs
- $\text{N}_2$ , CO,  $\text{CH}_4$  and  $\text{CO}_2$  desorption from dust analogues
- Summary

# Plasma generation of interstellar dust analogs



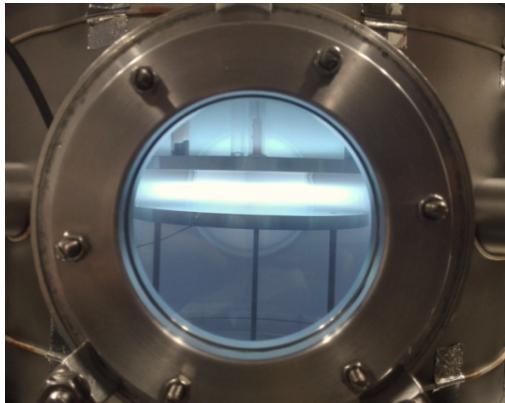
Inductive RF discharge

$\text{He}$ , 10 sccm  
 $\text{CH}_4$ , 5 sccm  
 $P_{\text{tot}}$ , 0.27 mbar  
40 W  
 $t_{\text{dep}}$ , 8 min  
Al substrate



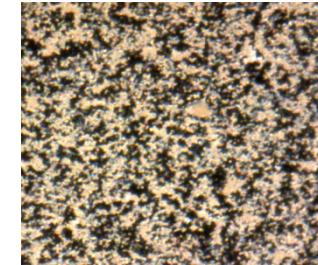
HAC1  
films, homogeneous

Peláez et al., 2018, Plasma Sources Sci. Technol.



Capacitive RF discharge

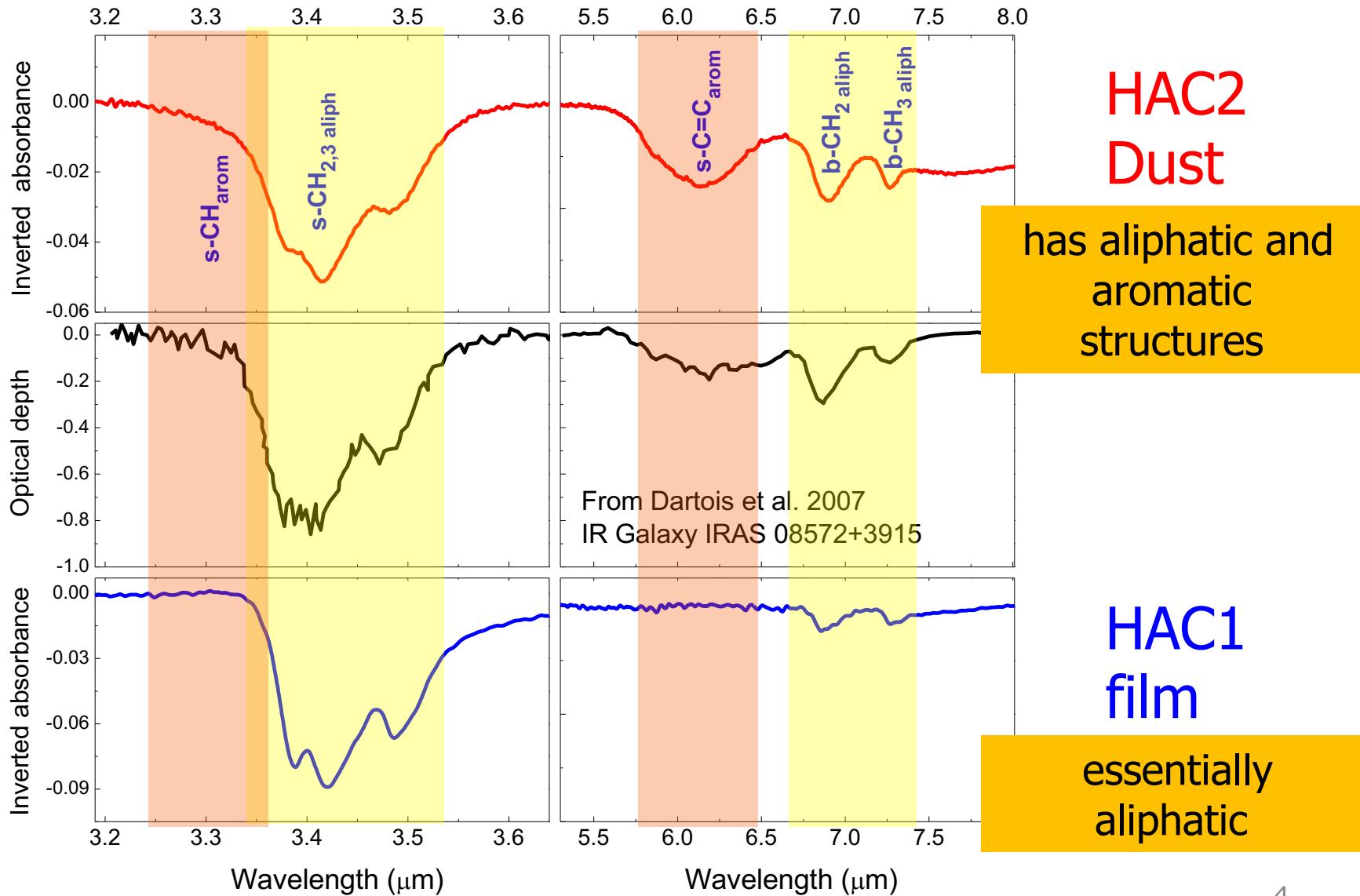
$\text{Ar}$ , 5 sccm  
 $\text{C}_2\text{H}_2$ , 2 sccm  
 $P_{\text{tot}}$ , 0.31 mbar  
15 W  
Modulation:  
15 s On/ 6 s Off  
 $t_{\text{dep}}$ , 80 min  
Al substrate



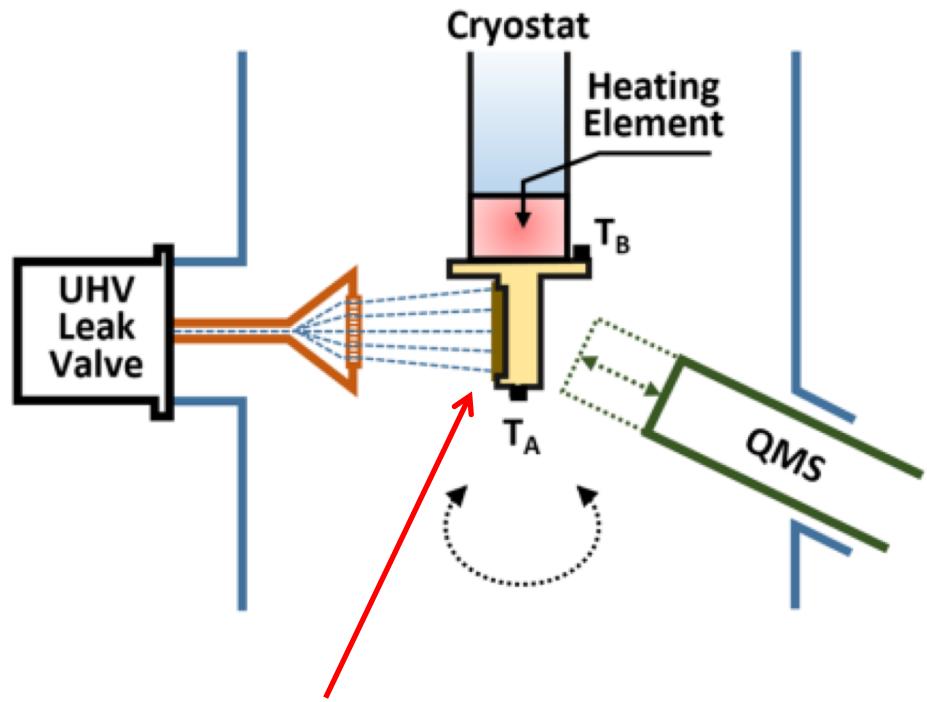
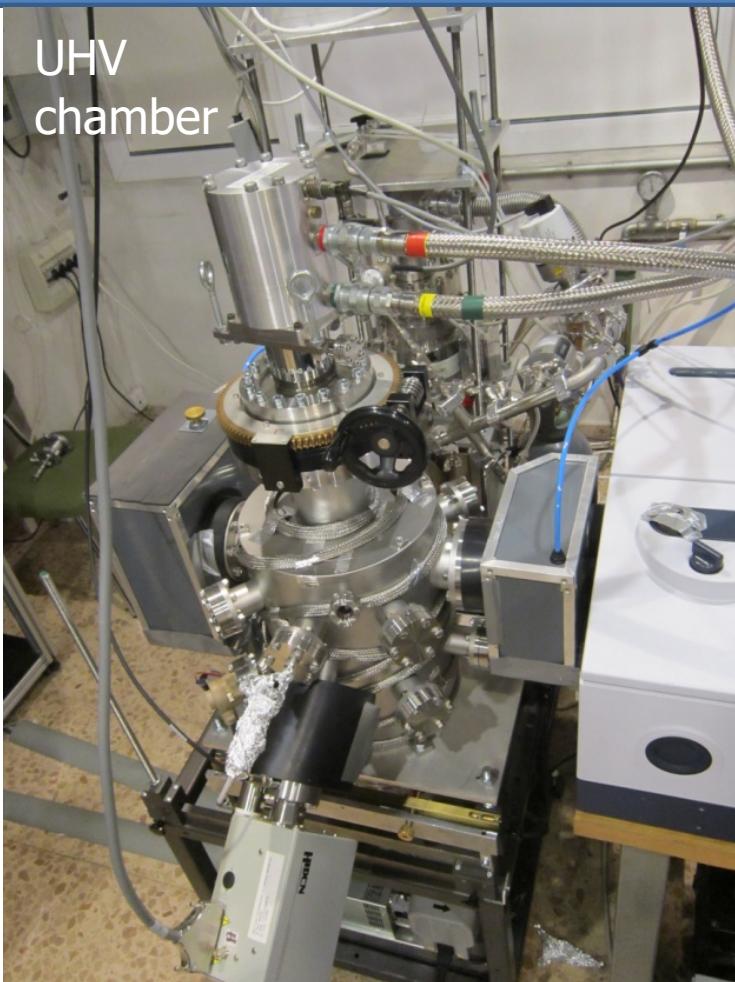
HAC2

SEM shows  
agglomeration of  
~150 nm diameter  
dust particles

# IR spectra: comparison with observations

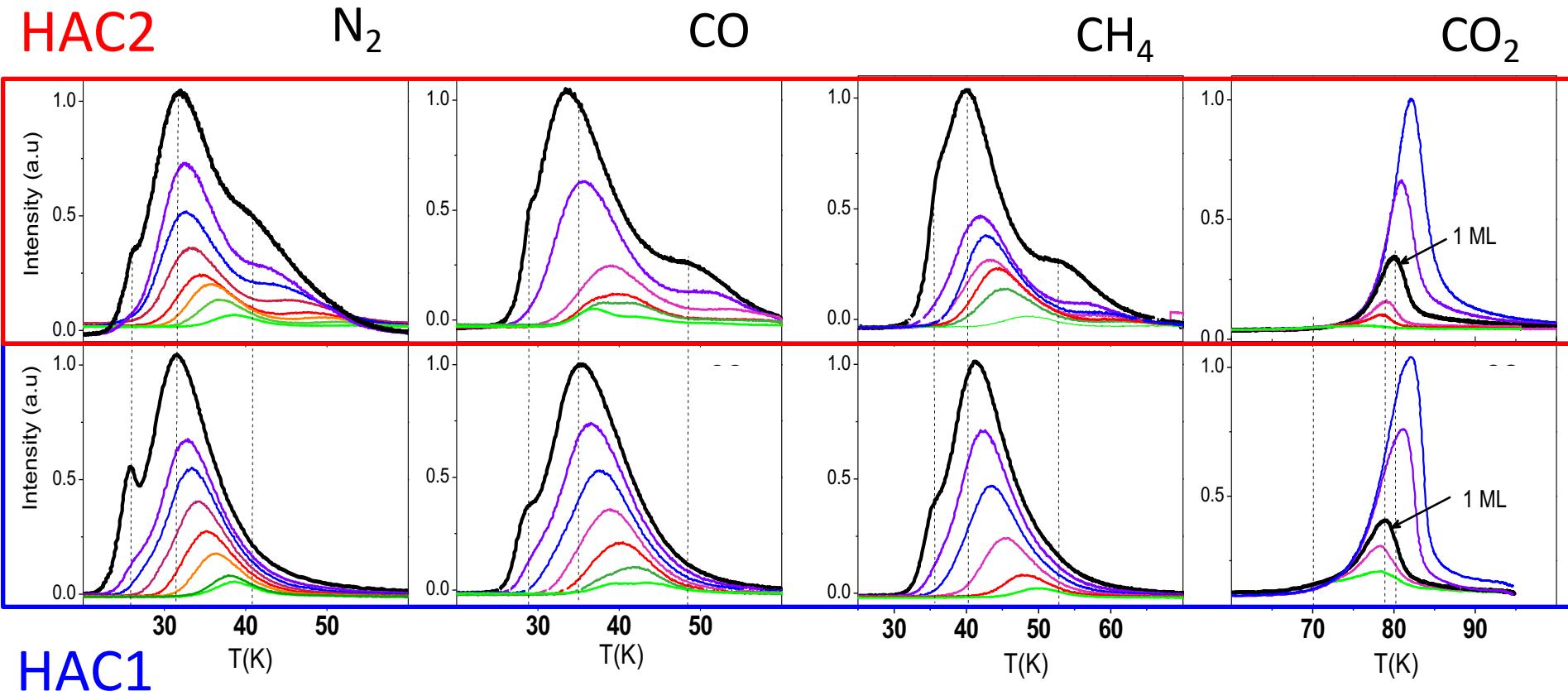


# Thermal programmed desorption of N<sub>2</sub>, CO, CH<sub>4</sub> and CO<sub>2</sub> from HAC



- Turbomolecular pump: S=700 l s<sup>-1</sup> for N<sub>2</sub>
- He closed cycle cryostat: 15-300 K
- Precision leak valve : down to 10<sup>-10</sup> mbar l s<sup>-1</sup>
- Base pressure, P<sub>b</sub> ≈ 1 10<sup>-10</sup> mbar

# TPD spectra of volatiles from HAC



- Desorption rate proportional to QMS signal  
( $\tau_{\text{res}} = (V/S) \approx 10 \text{ ms}$ )
- Heating ramp,  $\beta = 20 \text{ K/min}$
- $I_{\text{QMS}}(t) \rightarrow I_{\text{QMS}}(T)$

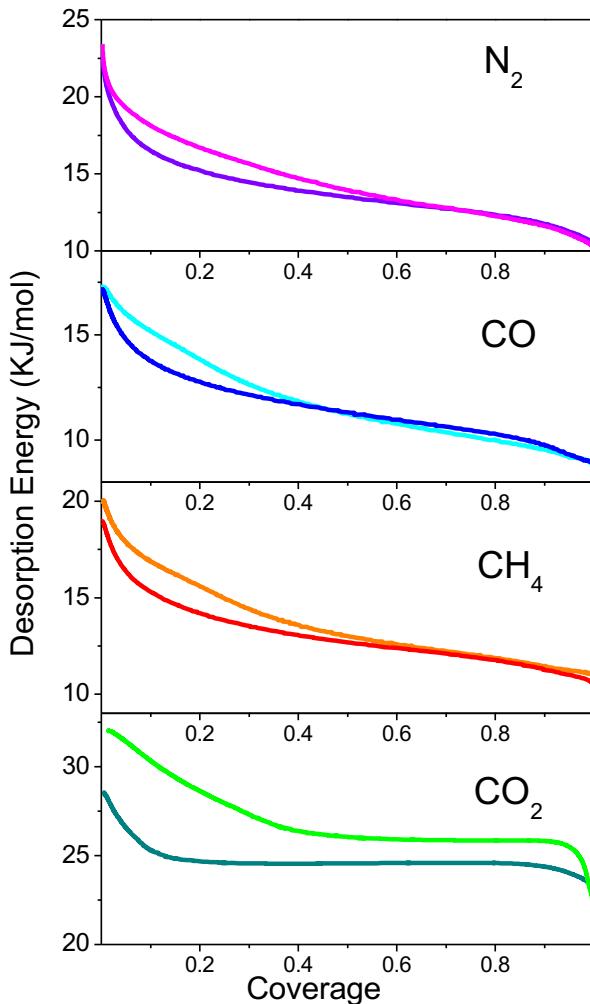
Study of sub-monolayer coverages,  
(black trace,  $\approx 1 \text{ ML}$ )

Maté et al., 2019 MNRAS, submitted

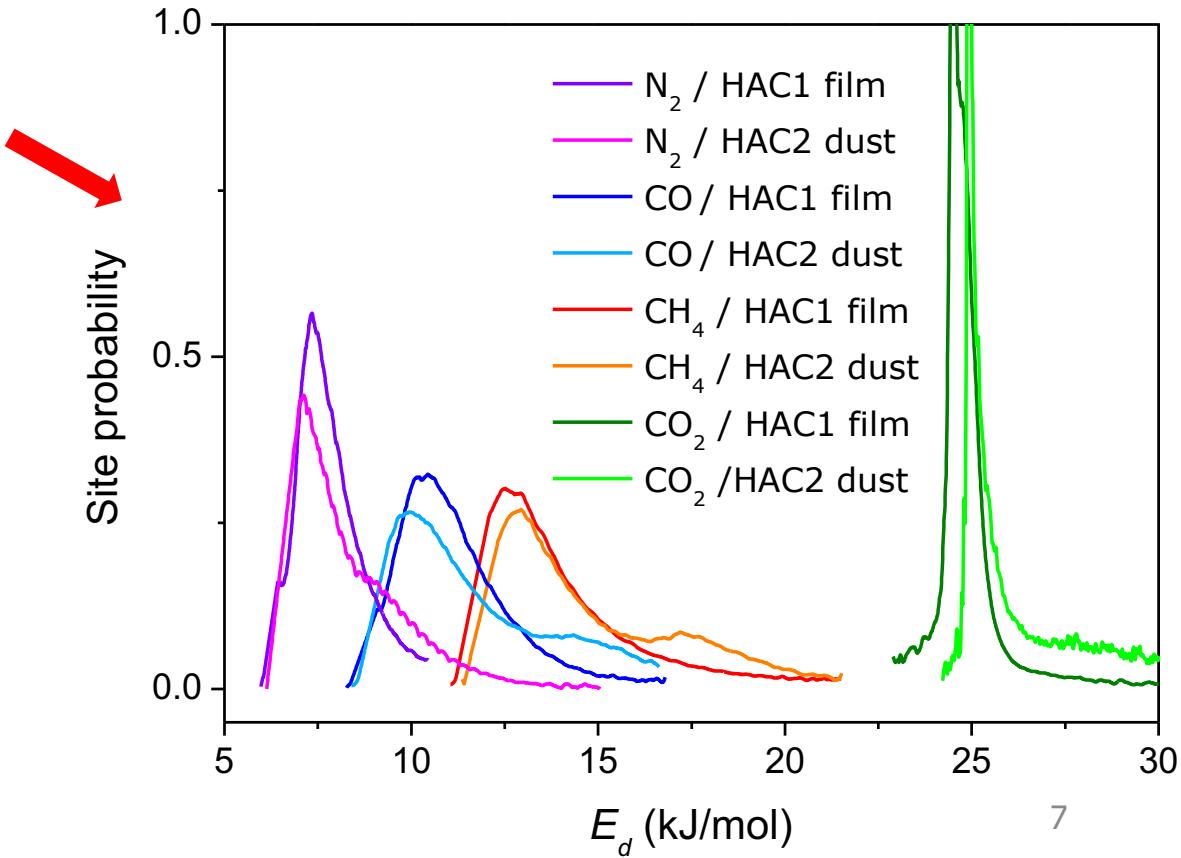
# Desorption energy: inversion analysis

**Polanyi-Wigner  
equation:**

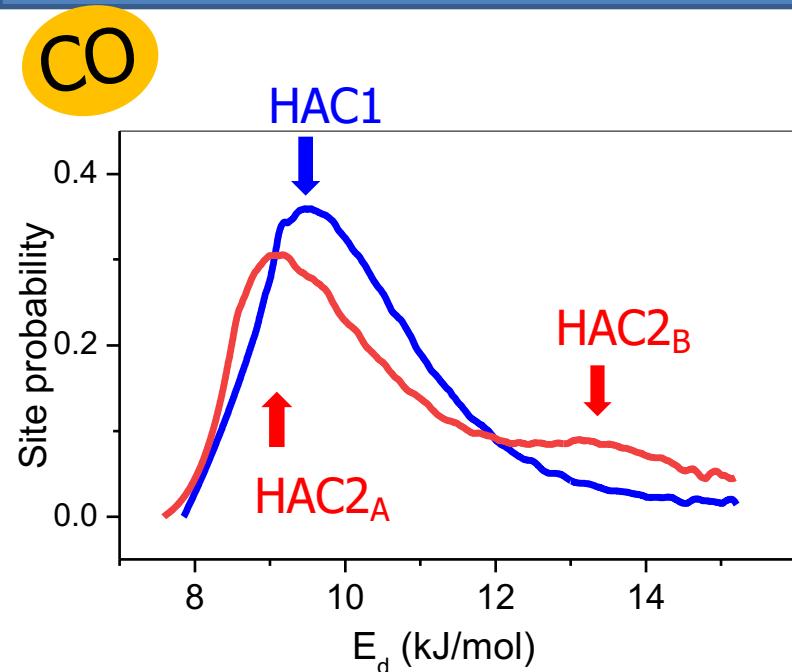
$$E_d(\theta) = -k_B T \ln\left(-\frac{d\theta/dt}{v\theta^n}\right)$$



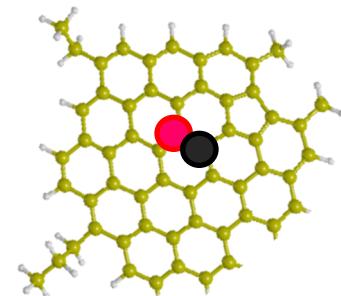
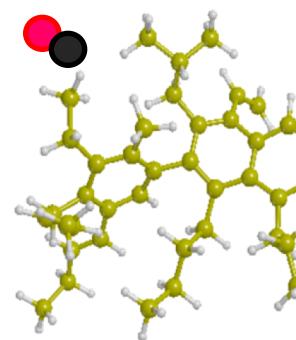
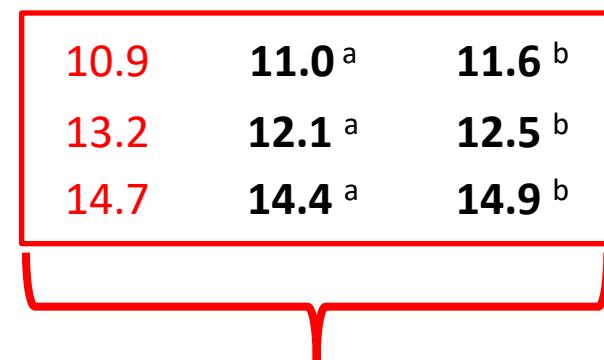
$E_d(\vartheta)$  = desorption energy  
 $v$  = pre-exponential factor :  $10^{13} \text{ s}^{-1}$ .  
 $\vartheta$  = coverage . Applied to  $\theta = 1$  (i.e. 1 ML)  
 $n$  = desorption order .  
 $n=1$  for N<sub>2</sub>, CO and CH<sub>4</sub>;  $n=0$  for CO<sub>2</sub>



# Aromatic and aliphatic surface sites



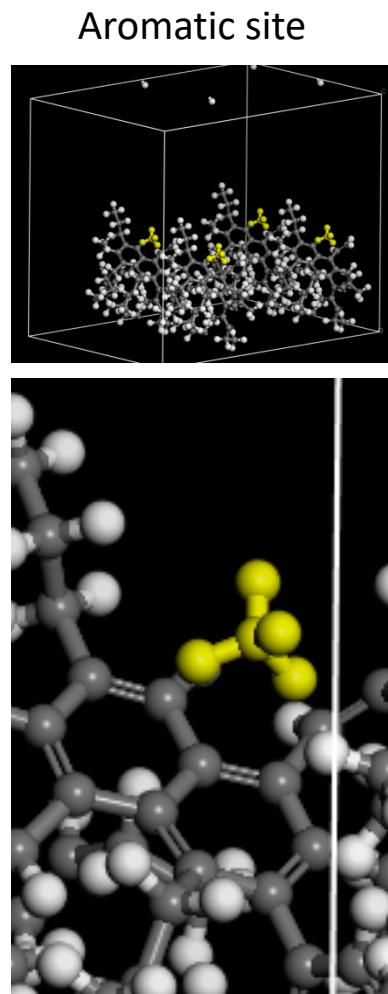
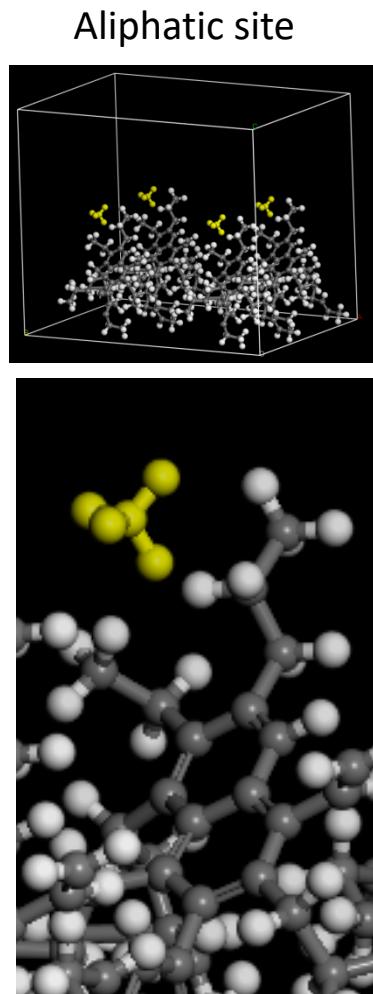
	$E_d$ (kJ/mol), for 1 ML				
	HAC1 film	HAC2 <sub>A</sub> dust	HAC2 <sub>B</sub> dust	Graphite	Graphene
N <sub>2</sub>	8.5	8.6	10.9	11.0 <sup>a</sup>	11.6 <sup>b</sup>
CO	9.6	9.1	13.2	12.1 <sup>a</sup>	12.5 <sup>b</sup>
CH <sub>4</sub>	11.2	10.8	14.7	14.4 <sup>a</sup>	14.9 <sup>b</sup>



a) H. Ulbricht et al. 2006 Carbon, 44, 2931. Redhead analysis, recalculated with  $v = 10^{13} \text{ s}^{-1}$

b) R. S. Smith et al. 2016, JPC B, 120, 1979. Inversion analysis. Average of energy distribution

# Aromatic and aliphatic surface sites: DFT calculations




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**EXPERIMENTAL,  $E_d$ (kJ/mol)**

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HAC1	HAC2 <sub>A</sub>	HAC2 <sub>B</sub>	graphite Ulbrich07
11.2	10.8	14.7	14.4

**CALCULATIONS,  $E$  (kJ/mol)  
(this work, preliminary)**

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aliphatic site	aromatic site	graphite
8.6	16.7	14.5

Surfaces from HAC solids described in Molpeceres et al. 2017, PCCP.

# Comparison with other astrophysical relevant surfaces

CO

Surface	Ed(kJ/mol) (pre-exp. factor)	Reference	Ed averaged
HAC1	<b><math>10.3 \pm 0.2 (10^{13} \text{ s}^{-1})</math></b>	this work	
HAC2	<b><math>10.6 \pm 0.2 (10^{13} \text{ s}^{-1})</math></b>	this work	
graphene	<b><math>12.5 (1.9 \times 10^{13} \text{ s}^{-1})</math></b>	Smith et al. 2016, JPC B	11.5
HOPG	<b><math>13 \pm 1 (6 \times 10^{14} \text{ s}^{-1})</math></b>	Ulbricht et al. 2006, C	
graphite	<b>11</b>	Vidali et al. 1991, SS	
SiO <sub>x</sub>	<b><math>7.2 (10^{12} \text{ s}^{-1})</math></b>	Noble et al. 2012, M	
SiO <sub>x</sub>	<b><math>7.3 \pm 0.2 (10^{12} \text{ s}^{-1})</math></b>	Collings et al. 2015, N	
Mg <sub>2</sub> SiO <sub>x</sub>	<b><math>9.3 \pm 0.1 (10^{12} \text{ s}^{-1})</math></b>	Suhasaria et al. 2017, A	
ASWcryst	<b><math>7.2 (10^{12} \text{ s}^{-1})</math></b>	Noble et al. 2012, M	
ASWnp	<b><math>8.4 (10^{12} \text{ s}^{-1})</math></b>	Noble et al. 2012, M	
ASWnp	<b><math>11.8 (3.5 \times 10^{16} \text{ s}^{-1})</math></b>	Smith et al. 2016, JPC B	
ASWnp	<b><math>8.1 (10^{12} \text{ s}^{-1})</math></b>	He et al. 2016, ApJ.	
ASWnp	<b><math>9.7 (7.1 \times 10^{11} \text{ s}^{-1})</math></b>	Fayolle et al. 2016, ApJL	
ASWp	<b><math>8.6 (10^{12} \text{ s}^{-1})</math></b>	He et al. 2016, ApJ.	8.9

 $E_d$  (carb. dust)

&gt;

 $E_d$  (ASW)

&gt;

 $E_d$  (silic. dust)

8.9

## Summary and conclusions

- Two HAC samples, taken as interstellar dust analogs, have been produced in plasma reactors. HAC1: film (mostly aliphatic structure). HAC2: agglomeration of dust particles (both aliphatic and aromatic structures).
- TPD experiments of N<sub>2</sub>, CO, CH<sub>4</sub> and CO<sub>2</sub> from the two carbonaceous solids were performed. The TPD spectra from the aliphatic film present just one maximum whereas those from the aliphatic/aromatic dust sample have a maximum and a shoulder.
- The comparison with literature values suggests **that the interaction of volatiles with the aromatic structures is stronger than with the aliphatic part of the solids.**
- In general, **interactions of volatiles with carbonaceous surfaces seem stronger than with other surfaces relevant in interstellar grains like water-ice or silicate analogs.**

# Thank you for your attention



<http://www.iem.cfmac.csic.es/fismol/fmap/>

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2.00um

SU8000 1.0kV 7.6mm x5.00k SE(UL) 4/9/2019 11:10

10.0um