



Astrochemistry of protoplanetary disks from an observational point of view

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Protoplanetary disks are the birthplace of planets



Study of the disk physical & chemical structure is fundamental to understand the formation of our **own solar system & extra-solar planetary systems**

Morphological diversity of disks

ALMA observations (mm grains)

SPHERE observations (µm grains)



Most of the disks display structures!



ALMA images: Fedele et al. (2018), van der Marel et al. (2016), Casassus et al. (2012) and Pérez et al. (2016)



SPHERE images: Garufi et al. (2017), van Boekel et al. (2017), Ginski et al. (2016), Pohl et al. (2017) and Stolker et al. (2016)

Planet Formation

Direct Imaging

SPHERE obs. (0.96–3.8 µm) of PDS 70 transition disk!



Kepler et al. (2018) Müller et al. (2018) Seen via gas and dust density perturbations

ALMA obs. (1.3mm) of AS 209



Disks are complex systems

Strong T & n gradients, UV & X-ray





"Inhomogeneous" chemistry
Gas/ice reservoirs are likely affected



Favre et al. (2013) Schwarz et al. (2016) Kama et al. (2016) Nomura et al. (2016) Miotello et al. (2017)

A depletion of elemental C in Solar Type (T Tauri) disks

CO abundance relative to H_2 : (0.1-3)x10⁻⁵ *in the disk's warm molecular layers* (*T*>20*K*), **lower than the canonical value of** χ (**CO**) = 10⁻⁴)

Carbon chemistry? (Aikawa et al. 1997, Reboussin et al. 2015)

CO chemical destruction via reactions with He+

Followed by rapid formation of carbon chains (C_xH_x) or CO_2

Freeze-out T higher than CO —> trap the carbon in ices

Carbon reservoir in gas? CO an unreliable tracer of the disk H₂ gas mass

Kastner et al. (2015) Bergin et al. (2016)



Molecular inventory of protoplanetary disks

Atoms			
C_{I}, C_{II}, O_{I}	Sturm et al. (2010), Meeus et al. (2012), Du et al. (2015), Kama et al. (2016)		
Ions			
HCO^+ , $H^{13}CO^+$, DCO^+ , N_2H^+ , CH^+	Dutrey et al. (1997, 2007), van Dishoeck et al. (2003), Thi et al. (2011)		
N_2D^+	Qi et al. (2008, 2013a), Öberg et al. (2015a), Huang & Öberg (2015)		
Carbon reservoirs?			
CO, CO_2	Koerner & Sargent (1995), Pontoppidan et al. (2010)		
Simple species			
¹³ CO, C ¹⁸ O, OH, HD	Dutrey et al. (1996), Pontoppidan et al. (2010),		
	Bergin et al. (2013), Favre et al. (2013), McClure et al. (2016)		
S-bearing molecules			
CS, SO, H_2S	Dutrey et al. (1997), Guilloteau et al. (2013), Phuong et al. (2018)		
13 CS, C 34 S, H ₂ CS	Le Gal et al. (2019)		
N-bearing molecules			
CN , HCN , $HC^{15}N$, $H^{13}CN$,	Dutrey et al. (1997), Qi et al. (2008), Salinas et al. (2016)		
HNC, DCN, NH ₃	Guzmán et al. (2017), Booth et al. (2019)		
Carbon chains			
$CCH, C_2H_2, c-C_3H_2, HC_3N$	Dutrey et al. (1997), Pontoppidan et al. (2010),		
	Henning et al. (2010), Chapillon et al. (2012), Qi et al. (2013b),		
	Oberg et al. (2015b),Bergner et al. (2018)		
Water			
H ₂ O	Bergin et al. (2010), Hogerheijde et al. (2011),		
	Fedele et al. (2012), Podio et al. (2013)		
O-bearing molecules			
H ₂ CO	Qi et al. (2013a),Loomis et al. (2015), Öberg et al. (2017)		
	Carney et al. (2017), Guzmán et al. (2018), Podio et al. (2019)		
t-HCOOH	Favre et al. (2018)		
Complex organic molecules			
CH ₃ OH	Walsh et al. (2016)		
CH ₃ CN	Oberg et al. (2015b),Bergner et al. (2018),Loomis et al. (2018)		

Infrared



Water reservoirs in disks

Planetesimal with H₂O form in the outer disk (water ice reservoir)

Courtesy Davide Fedele



Hot water chemistry & evaporation
Hot water chemistry
3 Photodesorption of water ice



Tielens & Hagen (1982): reaction scheme to form water ice



⁽¹ AU = Distance between the Sun and the Earth)

- Surface layers: molecules destroyed by UV photodissociation
- Inner (r <50 AU -100 AU, T >50-100K): molecules present in warm molecular layers. Production via gas phase chemistry or formation on ices and then release into the gas phase
- Outer disk / Mid plane (r >100-200 AU, T <50K): molecules are locked into the icy surface of dust grains (χ_{H2}~10⁻⁶-10⁻⁴), Only a few percent are in gas-phase (χ_{H2}~10⁻¹¹-10⁻⁷)

The chemical composition of disks is hidden in ices!

Interferometry is needed to access the molecular content in disk



High angular resolution!



50 AU = 0.4" at 140 pc

→ 100 AU = 0.7" at 140 pc

1 AU = Distance between the Sun and the Earth

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Interferometry is needed to access the molecular content in disk

The emissive area is expected to be small (and might be close to the central object) It is really hard to detect a not intense transition



Favre et al. (2013)

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Oberg et al. (2015a) Loomis et al. (2018)

Towards complex molecules in protoplanetary disks

Methanol (CH₃OH)

Key molecule in the formation routes to larger O-bearing molecules



T <u>ACKING</u>	Table 1Methanol Transitions	
Transition	Frequency (GHz)	Upper Level Energy (K)
$2_{11} - 2_{02}$ (A)	304.208	21.6
$3_{12} - 3_{03}$ (A)	305.473	28.6
$4_{13} - 4_{04}$ (A)	307.166	38.0
$8_{17} - 8_{08}$ (A)	318.319	98.8

Cold reservoir of CH₃OH in TW Hya non-thermal desorption (i.e. induced photo-desorption

and/or chemical desorption

Towards O-bearing molecules in protoplanetary disks



Podio,..., Favre et al. (2019) Qi et al. (2013) Carney et al. (2016) Öberg et al. (2017) Efficient formation of **organics on icy grains** for $R > R_{CO}$ and then desorption ?

Or gas phase chemistry?

Towards O-bearing molecules in protoplanetary disks

Formic acid (HCOOH)

Key molecule as the carboxyl group (C(=O)OH) is one of the main functional groups of amino acids



place at the verge of planet formation in protoplanetary disks

Favre, Fedele, Semenov et al. (2018)



Formation of planets via coagulation of dust grains:

1. *Dust* (μm)

Size & time

- 2. Pebbles (mm-cm)
- 3. Planetesimals (10-1000km)
- 4. Planets (10000km)



Courtesy C.Dominik

Summary

Water (warm and cold) & complex organic molecules (Nand O-bearing) are present in protoplanetary disks

Observations suggest that chemistry leading to molecular complexity likely takes place in protoplanetary disks where planets might form

- ISM inheritance?

- Reprocessed?

ALMA & NOEMA

(resolution and sensitivity)

key interferometers for astrochemical studies But still it will be difficult to detect larger species

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