### Today's PAH Model: Four Not So Easy Pieces

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# THANK YOU XANDER



~1985 - Backyard BBQ, San Jose



~1990 – Kuiper Airborne Observatory (KAO)



~1988 – 1<sup>st</sup> Ames Astrochem Lab Group



1988 - Interstellar Dust Proceedings IAU Symp 135

### Today's PAH Model: Four Not So Easy Pieces

## Outline

1. Spectroscopy between 1 -5  $\mu m$  and JWST

### 2. Formation and Growth

- 3. Destruction
- 4. Ice Chemistry

# PAH Spectroscopy between 1 -5 $\mu$ m and JWST<sup>4</sup>

The Near Infrared Spectrograph (NIRSpec) on *JWST* will measure spectra across the 1-5  $\mu$ m range with significantly greater sensitivity and resolving power than previously possible.

This opens up the study of weak, but important and unique, PAH transitions that are spread across this region

PAH Overtone, Combination, and Hot bands;

CD stretching bands in deuterated PAHs and

C=N stretching bands in PAH nitriles



Fig courtesy Els peeters

# PAH Spectroscopy between 1 -5 $\mu$ m and JWST<sup>2</sup>

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Vibration	Range in $\mu$ m
PAH CD stretch	4.3 - 4.5ª
dPAH CD stretch	4.54 - 4.75 <sup>b</sup>
PAH-nitrile C≡N stretch	4.46 - 4.5°

### JWST DISCOVERY SPACE

#### Weak PAH bands are expected in the $3.8 - 5\mu m$ region



wavelength (

AKARI spectra Doney et al. 2016, A&A, 586A,65D *ISO*, experimental, and computational spectra Peeters et al. 2004, ApJ 604, 252

### PAH-nitrile structures suggested by the detection of Benzonitrile (a) in TMC-1\*

Benzo-dinitrile (b), Benzo-trinitrile (c) and some pyrene-nitriles (d-f).



\*<u>McGuire</u> et al. *Science* 359, 202, 2018

Fig courtesy Christiaan Boersma

Fundamental and overtone band positions of the CD<sup>c</sup> stretch in deuterated PAHs and the C≡N stretch in PAH-nitriles between 1 and 5 µm.



Overtone band positions not corrected for anharmonicity

Fig courtesy Christiaan Boersma

### The NASA Ames PAH IR Spectroscopic Database

PAHdb

a Website, PAH spectroscopic library, IDL and Python suites



www.astrochem.org/pahdb

Boersma et al. 2013, ApJS 211, 8; Bauschlicher et al. 2018, ApJS 234, 32

### WHY PAHdb

Spectroscopic assignments are based on spectra of actual aromatic molecules in specific charge states, structures, sizes and so on. This allows the analysis of the spectra without the need of an ad-hoc interpretation of the state of the PAH population since the average synthesized spectra can be traced back to the fully characterized individual PAH molecules.

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IR spectra of late type stars (here CRL 618) show a wealth of lines from molecular intermediaries such as HC≡CH, HC≡N, HC≡C-C≡C-C≡N ..., the species that form carbon chains and PAHs



Cernicharo et al. 2001, ApJ 546, L123

Fig courtesy Christiaan Boersma

#### From carbon chains to rings





Carbon chains grow by sequential HC≡CH and HC≡N addition.

When long enough (~6-12 C atoms), chains cyclize.

Further addition of HC≡CH, HC≡N and their reactive forms lead to ring formation and growth.

C=N addition to  $H_2C=CH-CH=CH_2$ 

Morales et al. 2011, ApJ 742, 26

#### From small PAHs to larger PAHs



# Small PAHs grow to large PAHs by sequential HC≡CH and HC≡N addition and insertion.



Zhao, Kaiser & Xu 2018, Nature Astronomy

### From PAH clusters and large PAHs to particles and fullerenes..



#### Small cluster isomerization



Zhang et al. 2019, ApJ 872:38

Experiments showing possible steps toward particle and fullerene growth.

#### Large PAH fragmentation



Zhen et al. 2014, ApJ 797:L30

### **PAH Destruction**

PAH<sup>0</sup> UV DRIVEN FRAGMENTATION

H and  $H_2$  loss are the first steps

PAH edge shape influences odd to even H atom loss. H<sub>2</sub> loss dominates in larger PAHs

Ha HI H2 Ha HI Ha HI H2 Ha HI Ha HI H2 Ha HI HA H

Castellanos et al., 2018, A&A 616, A 166

PAH<sup>+</sup> UV DRIVEN FRAGMENTATION

Site selective  $C_2/C_2H_2$  loss and stepwise H loss

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UV driven fragmentation of substituted PAHs may contribute to formation of smaller species normally considered to form by merging atoms and molecules



Zhen et al. 2016, Molecular Astrophys 5, 1

### Putting it all Together

Schematic of PAH population changes with distance from the exciting star in the Reflection Nebula, NGC 2023.

PAH structures inferred from the fundamental PAH band groupings and ratios



Peeters et al. 2017, ApJ 836:198

Andrews et al. 2015, ApJ 807, 99

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#### PAH H<sub>2</sub>O Ice Chemistry

UV Photolysis studies of PAHs in low temperature ices show PAHs significantly alter the physical properties of the ice and chemical processes that occur within the ice.

• PAH ionization energy is lowered by 2 eV in H<sub>2</sub>O Ice

- PAH cations are easily produced and remain trapped up to ice temperatures near 100K
- Hydrogenated (H<sub>n</sub>PAHs) and oxidized PAHs (alcohols and ketones) are important photoproducts.

Bouwman et al. A&A 511, A33 (2010), Gudipati & Allamandola ApJ 596, L195 (2003) Bernstein, Sandford et al. Science 283, 1135 (1999) PAH/Ice photochemistry depends on ice temperature and PAH concentration

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- •Ions mediate the chemistry in ices below 50K, and at low concentrations :  $PAH/H_2O < 1/10^3$
- •Radicals mediate the chemistry in ices above 50K and at high concentrations:  $PAH/H_2O > 1/10^3$



Bouwman et al. ApJ (2009), Bouwman et al. A&A (2010) Cuylle et al. A&A 562, A22 (2014)

#### PAH/H<sub>2</sub>O Ice Photoproducts-

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# Cations (PAH<sup>+</sup>), Hydrogenated (H<sub>n</sub>PAHs) and Oxidized PAHs (alcohols and ketones)

#### Coronene and *identified* photoproducts



Guennoun, Aupetit, & Mascetti. 2011, PCCP 13, 730; Guennoun, Aupetit, & Mascetti. 2011, JPCA 115, 1844; Cook, Ricca, Mattioda et al. 2015, ApJ 799:14 (20pp); de Baros, Mattioda, Ricca et al. 2017, ApJ, 848, 112,

### Conclusions

- 1. JWST's unprecedented sensitivity, spectroscopic resolution and bandwidth will open the 1-5  $\mu m$  region
- 2. The experimental, theoretical and computational tools that have been developed over the past few years are revealing a detailed and surprising picture of how PAHs grow and evolve.
  - This information will guide model development, clarify, and enable quantification of the many roles PAHs play in astrophysics.
- 3. PAH/ice chemistry and spectroscopy should be considered in astrochemical models.



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