The atomic to molecular Hydrogen transition: A major step in the understanding of PDRs.

Introduction

The multiple facets of molecular Hydrogen a new picture of PDRs provided Herschel observations Conclusion and perspectives



- \* Photodesorption
- Conversion of UV \* to IR photons



IR – submm – mm : discrete transitions

Chemistry : 
$$\frac{d[X]}{dt} = F - D[X] = 0$$

Gas phase: reaction rate coefficients are radiation field and temperature dependent

Dust and gas/solid interface: at least for H<sub>2</sub> formation

#### Thermal balance : $\Delta G = H - C = 0$

Heating: photoelectric effect, collisional de-excitation of pumped H<sub>2</sub>, cosmic rays, chemical formation through exothermic reactions

Cooling: radiative emission following collisional excitation

kinetic energy => radiative energy Solve the detailed balance (collisional excitation, radiative emission/pumping) endothermic chemical reactions Gas-grain coupling



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# Inputs /outputs



The Meudon PDR code @ <u>http://ism.obspm.fr</u>

# Inputs /outputs



Benchmark exercise : Röllig et al. 2007, A&A 467, 187

## Physics of $H_2$ + photons with hv < 13.6eV



by Wolniewicz et al. 1998, ApJS 115, 293

# Full infrared spectrum of H<sub>2</sub>

Electric quadrupole ( $\Delta J = 0, \pm 2$ ) + magnetic dipole ( $\Delta J = 0$ ) transitions available within X <sup>1</sup> $\Sigma$ <sup>+</sup> ground electronic state (Pachucki + Komasa 2011, PRA 83, 032501) (Roueff et al 2019, A&A, in press)

302 bound levels (with v=14, J=4)

4711 transitions

10<sup>-3</sup> to 10<sup>-4</sup> cm<sup>-1</sup> wavenumber accuracy which translates to  $\approx 10^{-7}$  - 10<sup>-8</sup> µ for the wavelengths (Pachucki et al. submitted to PRA).

 $\Delta v=1$ , excited J transitions have larger MD emission probabilities than EQ values.



being included in Meudon PDR code to be updated for future JWST searches



# modelling of interstellar H<sub>2</sub>

### Statistical equilibrium of v, J levels within X 14

- **\*** collisions with H, He, H<sup>+</sup>, e, H<sub>2</sub>, ...
- \* radiative cascades
- v, J dependent chemistry (gas phase + surface)

### UV pumping (radiative transfer)

Line self-shielding + dust shielding

### Coupling to cooling / heating

- Formation from H + H on grain surfaces releases ~ 4.5eV to be shared between H<sub>2</sub> internal, kinetic and grain energies
- Collisions with H, He, ... may lead to cooling if  $A_{v'J'->v",J"} > k^{H}_{v'J'->v",J"} n(H) + k^{He}_{v'J'->v",J"} n(He) + ... where kinetic$ energy from excitation collisions istransferred into radiative energy orheating if collisional desexcitation >radiative quenching



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$$\frac{dn(\mathbf{H}_2)}{dt} = \frac{1}{2}R_{\mathbf{H}}n(\mathbf{H})$$
$$R_{\mathbf{H}} = s < \sigma n_d > \bar{v}_{\mathbf{H}}$$

 $< \sigma n_d >$  is the mean cross section of dust grains per unit volume

Assuming identical spherical dust grains of radius a

$$\langle \sigma n_d \rangle = \frac{3}{4} \times \frac{1.4m_{\rm H}G}{\rho_d} \times \frac{1}{a} \times n_{\rm H}$$

With G : dust to gas mass ratio  $\approx 10^{-2}$   $\rho_d$  : volumic mass of dust  $\approx 3~g~cm^{-3}$  a = 0.1  $\mu m$ 

$$R_{\rm H} = 1.6 \times 10^{-16} \times s(T_d, T_K) \times \sqrt{\frac{T_K}{300}}$$

 $\approx$  Tielens & Hollenbach 1985, ApJ 291: 722

 $s = \text{sticking coefficient} \le 1$ , depends on T<sub>d</sub> and T<sub>K</sub> (Hollenbach & McKee 1979)

With the Mathis et al. (1977) dust grain size power law distribution  $\approx a^{-3.5}$ ,

$$<\sigma n_d>=\frac{3}{4}\times\frac{1.4m_{\rm H}G}{\rho_d}\times\frac{1}{\sqrt{a_{min}\cdot a_{max}}}\times n_{\rm H} \longrightarrow R_{\rm H} = 4.8\times10^{-16}\times s(T_d,T_K)\times\sqrt{\frac{T_K}{300}}$$

with  $a_{min} = 3 \text{ nm}$ ,  $a_{max} = 0.3 \mu$ 

# The chemical phases of Photodissociation regions









#### Langmuir-Hinshelwood / Physisorption

- ♦ binding energy  $\approx 658$  K
- diffusion energy  $\approx 510$  K
  - → Formation of H<sub>2</sub> is efficient only in a narrow range of dust temperatures

#### **Eley-Rideal / Chemisorption**

 binding energy: several thousands K efficient at the edge of PDRs
 Cazaux & Tielens, ApJ575, L29, 2002



#### Langmuir-Hinshelwood / Physisorption

- ♦ binding energy  $\approx 658$  K
- diffusion energy  $\approx 510$  K
  - → role of temperature fluctuations on small grains help (Bron et al. 2014, A&A 569, A100)

#### **Eley-Rideal / Chemisorption**

 binding energy: several thousands K efficient at the edge of PDRs chemisorption threshold : 300 K (to reach a chemisorption site)

The efficiency of the **LH** mechanism depends on **grain** temperatures The efficiency of the **ER** mechanism depends on the **gas temperature** 



when  $H_2$  heats the gas  $\rightarrow$  positive loop to form  $H_2$  and heat the gas

# Heating by H<sub>2</sub>



For high densities, H<sub>2</sub> may convert UV photon energy into kinetic energy





### **Two archetypal PDRs**

Herschel GTKPs observations of two nearby, bright galactic PDRs : Orion Bar (Hexos, PI T. Bergin), and NGC 7023 NW (WADI, PI V. Ossenkopf-Okada)





Rotationally excited CO

**\*** In addition to HIFI, SPIRE, PACS Herschel data:

observations from Spitzer, ISO, CFHT, IRAM-PdB, ...

**Observational constraints** 

Full <sup>12</sup>CO ladders (4-3 to ~ 20-19), H<sub>2</sub> ro-vibrational lines, <sup>13</sup>CO, CH<sup>+</sup>, HCO<sup>+</sup>, C<sup>+</sup>, O, HD

Total: ~ 40 lines

Astrophysical knowledge

Properties of illuminating stars / clusters is known  $\rightarrow$  UV field intensity is known a priori NGC 7023 NW: G0 ~ 2600 Orion Bar: G0 ~ 2×10<sup>4</sup>

Modelling: difficult (impossible) to reproduce CO excitation in constant density models

→ isobaric PDR models

### Two archetypal PDRs

## A new picture of PDRs

use of a grid of isobaric PDR models

Only 2 free parameters:

- thermal pressure
- beam filling/geometrical factor

Allows to consistently satisfy the constraints:

~20 constraints used in the fit

(+6 other successfully reproduced a posteriori)

Excited CO, H<sub>2</sub>, CH<sup>+</sup>, HD, HCO<sup>+</sup>, ...

#### ➡ High thermal pressures ~10<sup>8</sup> K cm<sup>-3</sup>

but does not reproduce tracers in the atomic/H II region or deep inside the cloud

Warm compressed layer at the edge of PDRs



## far-reaching effects of H<sub>2</sub> formation

effect of the chemisorption threshold

Stronger impact on excited CO lines than on H<sub>2</sub> lines !

Controls the amount of warm molecular gas (1000-100 K)  $\rightarrow$  new active chemistry



## far-reaching effects of H<sub>2</sub> formation



⇒ surmounting activation barriers:  $O + H_2(v,J) \Rightarrow OH + H$  (3241 K)  $OH + H_2(v,J) \Rightarrow H_2O + H$  (1751 K)

### far-reaching effects of H<sub>2</sub> formation



~ 50% of CO: 19-18 emission comes from a region in front of the C/CO transition

## warm chemistry at work at the edge

Formation of H<sub>2</sub> occurs in a region:

- high temperature
- UV photons (thanks to an efficient formation rate)





chemical reaction rate dependent on H<sub>2</sub>(v,J): state-dependent chemistry

H<sub>2</sub> internal energy used to overcome endothermicity / activation thresholds warm chemistry at the edge of PDRs leads to the formation of CO via OH and CH<sup>+</sup>

### relation between p and UV intensity

#### Comparison to other PDRs



The UV radiation field seems to be responsible for the high pressures at the edge of PDRs

Goicoechea et al. (2016, Nature 537, 207)



- UV photons can bring the energy to excite CO in PDRs
- no need of additional hypothesis (additional heating sources, clumps, ...)
- the molecular edge of PDRs is characterized by high pressures maintained dynamically

Link between UV field intensity and gas pressure at the edge of PDRs found in other lines of sight (eg. Trumpler 14 in Carina)



Wu, Bron et al. 2018, A&A 618, A53

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# Emphasis on $H_2$

Introduction of detailed H<sub>2</sub> formation/excitation/destruction mechanisms

under high density and high radiation field conditions, radiative energy can be converted into kinetic energy

efficient heating at the edge of PDRs

- $\implies$  positive loop to form H<sub>2</sub> and heat the gas
- $\blacksquare$  H<sub>2</sub> state dependent chemistry at work (reactions with C<sup>+</sup>, S<sup>+</sup>, O, C, N ...)

Far reaching effects of H<sub>2</sub>

- CO formation channel through CH<sup>+</sup> + O, OH + C<sup>+</sup>
- excited CO transitions are emitted before the C+/C/CO transition
- no additional hypothesis (clumps, shocks, ...) required

Link between the strength of the impinging radiation field and gas pressure applicable from diffuse/translucent clouds to strong PDRs

#### For the future: observations

another promising source: Herschel 36 in the bright Messier 8 face-on PDR (Rachford et al. 2014, ApJ 786, 159, Oka et al. 2013, ApJ 773, 42, Tiwari et al. 2019, A&A 626, A28)
→ talk by Tiwari at 16:55 : link to hydrocarbon chemistry

preparing JWST (spatial resolution ≈ 10<sup>-3</sup> pc in the Orion bar) allows to resolve the hot layer of PDRs JWST Early Release Science program: radiative feedback from massive stars (PIs: O. Berné, E. Habart, E. Peeters)

Access to main key processes in PDRs

- H<sub>2</sub> formation rate
- Photo-electric heating & H<sub>2</sub> heating rates
- warm chemistry

### For the future: modelling efforts

- → introduce PAHs, photo-desorption and more complex surface chemistry talks of C. Joblin, P. Bréchignac, F. Dulieu, S. Cazaux, …
- → new class of models

UV photons can induce compression through photoevaporation



- Extend the studies for small clumps inside HII regions (e.g. Bertoldi & McKee 1990) to the edge of larger scale molecular clouds (Bron et al. 2018, arxiV, Kirsanova & Wiebe 2019, MNRAS 486, 2525)
  - $\rightarrow$  impact on the PDR structure ?

### For the future: debating with Xander on the formation of H<sub>2</sub>

Eley-Rideal / Photoinduced polycyclic aromatic hydrocarbon dehydrogenation. (Castellanos et al. A&A616, A167, 2018)

### and specific thanks to my young colleagues

Emeric Bron, Franck Le Petit, Jacques Le Bourlot







# Stochastic processes

New formalism to simulate H<sub>2</sub> formation

- Master equation formalism to consider fluctuations of the dust temperature and of the surface chemical state
- PDF of dust temperature + populations of adsorbed species on grains
- Mean formation rate coupled to the Meudon PDR code





**Observations ISO/Spitzer (PDRs)** 

- Efficient formation on small grains
- Grains spend most time being cold
- Langmuir-Hinshelwood alone could be sufficient in most PDRs